Efficient Constructionof4-aryl-5,8-epiminobenzo[7]annulenesbased on a Cascade Michael/Cyclization Reaction

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Summary:A practical method was developed forthegain of potentially biologically active 4-aryl-5,8-epiminobenzo[7]annulenesusingtropinoneas starting material withanazabicyclo[3.2.1]octan skeleton.In an effort to improve product yield, reaction process conditions were optimized and the cascade Michael/cyclization reaction went most smoothly using tetrahydrofuran as solvent in the presence of DBU at 60℃ for 10 hours. More diverse 4-aryl-5,8-epiminobenzo[7]annulenes were synthesized in good yields and structurally identified by NMR, FTIRand mass spectrometry analysis.The assembly of the heterocyclic core proceeds by a cascade Knoevenagel condensation, Michael addition and cyclocondensation sequence with a broad substrate applicability and good functional group tolerance.

Keywords:Tropinone; 5,8-epiminobenzo[7]annulenes; DBU; Michael addition; Cyclization

Introduction

The 5,8-epiminobenzo[7]annuleneskeletonis widely distributed in natural products aswellas synthetic compoundswith various valuablebiological activity, such as anticonvulsant[1-2], *N*-methyl-*D*-aspartate (NMDA) antagonist [3-5], phenylethanolamine*N*-methyltransferase (PNMT) inhibitor [6], antitumor [7-8] and treatmentoftype 2 diabetes[9].Simultaneously, the compound contains a 5,8-epiminobenzo[7]annulene skeleton can be used as a raw material for the synthesis of other useful compounds [10-12].The medicinal relevance and other important applications of 5,8-epiminobenzo[7]annulenederivatives haveattracted more and more attention among synthetic chemists, and a lot ofdifferent synthesis methods wereexplored and proposed for the gain of 5,8-epiminobenzo[7]annuleneframeworks.In its characteristic molecular structure, a pyrrolidine ring was essentially contained.As an efficient way for the assemblyof pyrrolidine ring, 1,3-dipolar cycloaddition with azomethine ylidescan be employed for the design of 5,8-epiminobenzo[7]annulenes, such as asymmetric [3+2]-dipolar cycloaddition [13-14], cycloisomerization/dipolar cycloaddition [15-16], dehydrogenative [3+2] cycloaddition [17] and intramolecular cross[3+2] cycloaddition [18]. In addition, a fewdifferentsynthesis strategies, includingradical translocation/cyclization [19], cyclization[20], intramolecular alkene carboamination [21], intramolecular ringclosure [22], tandem C-H amination [23], hydroamination [24], and formal carbenoidinsertion into the C−N bond in amide [25], have been developed for the construction of 5,8-epiminobenzo[7]annulene skeleton. Although rich and diverse synthetic strategies have been explored,it is stilla great challenge for the proposal of more novel methods with high efficiency and operability.

Recently, tropinone has been taken as a structural core for the gain of various tropinone derivatives. Some of these compounds were obtained through the reaction of tropinone and aromatic aldehydes, such as 2,4-di((*E*)-arylidene)-8-azabicyclo[3.2.1]octan-3-ones[26-28], 2-((*E*)-arylidene)-8-azabicyclo[3.2.1]octan-3-ones [29-30], and 2-(hydroxy(aryl)methyl)-8-azabicyclo[3.2.1]octan-3-ones [31-33].Moreover, these compoundscan be further converted to other heterocyclic compounds [34-36]. In addition, 2-(8-azabicyclo[3.2.1]octan-3-ylidene)malononitriles can be synthesized by the reaction between tropinone analogs and malononitrile, and can be used as a raw material in many other reactions [37].However, there was few research about the direct synthesis of 4-aryl-5,8-epiminobenzo[7]annulenes through one-pot reaction of tropinone, malononitrile and aldehydes. As a classical methodology, the Michael addition was a powerfultool for constructing new carbon–carbon bond playing a key role inmanymulticomponent reactions (MCR) especially in the construction of novel ring,which convertedatleast three different and easily accessible starting materials to the expected

productsin only one step with reduced consumption, increased output and simplified operation[38-40]. Therefore, we try hard to supplement a different and effectivemethodfor the preparation of a series of4-aryl-5,8-epiminobenzo[7]annulenesby a tandem Michael addition−Cyclization reaction with acceptable results.

Experimental

All chemical reagents needed in the experiment were purchased from some reliable reagent companiesandusedas received.Thin layer chromatography (TLC) was used forthe analysis ofallreaction process, and column chromatography was applied to separate targetcompounds.The NMR spectra with the reported chemical shifts in ppm were applied forthe structure characterization ofall target compounds,and sample test was carried outon a Bruker AM400 NMR spectrometer utilizingtetramethylsilane (TMS) as the internal standard. The IR spectra of all products were determined by a ThermoFisher FTIR spectrometer. Negative ion TOF-MS data of all compounds wereacquiredfrom an Agilent mass spectrometer.

General experiment process and operation steps for the one-potsynthesis of 4-aryl-5,8-epiminobenzo[7]annulenes **4**:Tropinone**1** (0.3 mmol), malononitrile **2** (1.2 mmol), aromatic aldehydes**3**(0.3 mmol) and DBU (0.6 mmol) were taken and added to dry glass tubes equipped with a stirring bar, and then 3mLtetrahydrofuran was added to dissolve the reactants. All reaction tubes were placed in a constant-temperature oil bath,the reaction temperature was set at60°C, and the reaction solutionwas stirred for 10 hours. Subsequently, the desired product was isolated by column chromatography utilizing petroleum ether/ethyl acetate (1:1, v/v) as eluent. Finally, target product was concentratedmaking the use of a rotary evaporator and the residual organic solvent was removedbya vacuum drying oven. The spectral data of purecompoundsislistedas follows.

Compound**4a**: Yellow solid; mp: 272-273°C.^{1}H NMR ($CDCl_3$): δ =7.69-7.08 (m, 4H, Ar-H), 5.10 (s, 2H, NH_2), 3.59-3.57 (m, 1H), 3.53-3.50 (m, 1H), 3.36-3.27 (m, 1H), 2.69-2.64 (m, 1H), 2.27 (s, 3H, CH_3), 2.18-2.09 (m, 2H, CH_2), 1.74-1.57 (m, 2H, CH_2).IR (KBr): 3382, 3117, 2922, 2849, 2796, 2213, 1566, 1490, 1449, 1296, 1266, 1254, 1236, 1159, 1140, 1071, 1011, 826, 787, 754cm^{-1}.MS-ESI: *m/z* 391.06 [M-H]$^-$.

Compound**4b**: Yellow solid; mp: 296-298°C. ^{1}H NMR ($CDCl_3$): δ =7.16-7.10 (m, 4H, Ar-H), 4.99 (s, 2H, NH_2), 3.51-3.49 (m, 1H), 3.43-3.41 (m, 1H), 3.26-3.20 (m, 1H), 2.60-2.55 (m, 1H), 2.18 (s, 3H, CH_3), 2.15-2.03 (m, 2H, CH_2), 1.66-1.61 (m, 2H, CH_2). IR (KBr): 3387, 3238, 2923, 2849, 2797, 2217, 1606, 1564, 1512, 1484, 1454, 1296, 1258, 1231, 1158, 834, 787, 752cm^{-1}.MS-ESI: *m/z* 331.14 [M-H]$^-$.

Compound**4c**: Yellow solid; mp: 275-277°C. ^{1}H NMR ($CDCl_3$): δ = 7.44-7.05 (m, 4H, Ar-H), 5.02 (s, 2H, NH_2), 3.49-3.48 (m, 1H), 3.43-3.40 (m, 1H), 3.26-3.20 (m, 1H), 2.59-2.55 (m, 1H), 2.17 (s, 3H, CH_3), 2.14-2.01 (m, 2H, CH_2), 1.70-1.48 (m, 2H, CH_2). IR (KBr): 3382, 3237, 3124, 2919, 2848, 2797, 2214, 1567, 1494, 1451, 1296, 1255, 1237, 1159, 1141, 1092, 1014, 829, 791, 756cm^{-1}.MS-ESI: *m/z* 347.11 [M-H]$^-$.

Compound**4d**: Pale yellow solid; mp: 278-280°C. ^{1}H NMR ($CDCl_3$): δ = 7.47-7.07 (m, 4H, Ar-H), 5.06 (s, 2H, NH_2), 3.57-3.47 (m, 2H), 3.33-3.28 (m, 1H), 2.67-2.61 (m, 1H), 2.25 (3H, CH_3), 2.22-2.07 (m, 2H, CH_2), 1.73-1.63 (m, 2H, CH_2). IR (KBr): 3387, 3312, 3122, 2924, 2849, 2794, 2211, 1564, 1452, 1354, 1296, 1255, 1237, 1162, 1140, 1079, 927, 887, 780, 751, 717cm^{-1}.MS-ESI: *m/z* 347.11 [M-H]$^-$.

Compound**4e**: Black solid; mp: 288-290°C. ^{1}H NMR ($CDCl_3$): δ = 8.05-7.33 (m, 4H, Ar-H), 5.03 (s, 2H, NH_2), 3.45-3.37 (m, 2H), 3.28-3.22 (m, 1H), 3.09 (s, 3H, CH_3), 2.63-2.58 (m, 1H), 2.18 (s, 3H, CH_3), 2.15-2.02 (m, 2H, CH_2), 1.65-1.61 (m, 2H, CH_2). IR (KBr): 3359, 3252, 3065, 2923, 2870, 2211, 1568, 1493, 1450, 1400, 1303, 1150, 1087, 1062, 960, 766, 545cm^{-1}.MS-ESI: *m/z* 391.13 [M-H]$^-$.

Compound**4f**: Tawny solid; mp: 270-272°C. ^{1}H NMR ($CDCl_3$): δ = 7.73-7.25 (m, 4H, Ar-H), 5.04 (s, 2H, NH_2), 3.44-3.42 (m, 2H), 3.28-3.22 (m, 1H), 2.61-2.57 (m, 1H), 2.18 (s, 3H, CH_3), 2.08-1.97 (m, 2H, CH_2), 1.66-1.61 (m, 2H, CH_2). IR (KBr): 3342, 3233, 3061, 2953, 2920, 2850, 2801, 2223, 1570, 1467, 1406, 1331, 1287, 1270, 1167, 1124, 1106, 1068, 1023, 864, 840cm^{-1}.MS-ESI: *m/z* 381.14 [M-H]$^-$.

Compound**4g**: Pale yellow solid; mp: 191-193°C. ^{1}H NMR ($CDCl_3$): δ = 7.77-7.39 (m, 4H, Ar-H), 5.09 (s, 2H, NH_2), 3.52-3.46 (m, 2H), 3.35-3.29 (m, 1H), 2.69-2.63 (m, 1H), 2.25 (3H, CH_3), 2.20-2.02 (m, 2H, CH_2), 1.79-1.70 (m, 2H, CH_2). IR

(KBr): 3373, 3226, 2925, 2853, 2216, 1571, 1449, 1354, 1326, 1308, 1270, 1168, 1122, 1073, 809, 703cm^{-1}.MS-ESI: *m/z* 381.14 [M-H]$^{-}$.

Compound**4h**: Yellow solid; mp: 249-250°C. ^{1}H NMR ($CDCl_3$): δ =7.84-7.32 (m, 4H, Ar-H), 5.11 (s, 2H, NH_2), 3.52-3.45 (m, 2H), 3.34-3.30 (m, 1H), 2.68-2.64 (m, 1H), 2.25 (s, 3H, CH_3), 2.22-2.08 (m, 2H, CH_2), 1.69-1.56 (m, 2H, CH_2). IR (KBr): 3340, 3233, 2921, 2852, 2230, 1568, 1511, 1455, 1351, 1287, 1263, 1167, 1136, 1107, 1071, 868, 845cm^{-1}.MS-ESI: *m/z* 338.15 [M-H]$^{-}$.

Compound**4i**: Yellow solid; mp: 297-298°C. ^{1}H NMR ($CDCl_3$): δ =8.33-7.32 (m, 4H, Ar-H), 5.05 (s, 2H, NH_2), 3.45-3.39 (m, 2H), 3.28-3.23 (m, 1H), 2.63-2.58 (m, 1H), 2.18 (s, 3H, CH_3), 2.15-2.03 (m, 2H, CH_2), 1.64-1.57 (m, 2H, CH_2). IR (KBr): 3352, 3241, 2922, 2850, 2218, 1565, 1517, 1454, 1344, 1264, 1104, 1016, 850, 733, 702cm^{-1}.MS-ESI: *m/z* 358.14 [M-H]$^{-}$.

Compound**4j**: Tawny fawn solid; mp: 261-262°C. ^{1}H NMR ($CDCl_3$): δ =8.30-7.46 (m, 4H, Ar-H), 5.07 (s, 2H, NH_2), 3.45-3.38 (m, 2H), 3.28-3.23 (m, 1H), 2.64-2.56 (m, 1H), 2.19 (3H, CH_3), 2.12-1.99 (m, 2H, CH_2), 1.73-1.63 (m, 2H, CH_2). IR (KBr): 3384, 3181, 2924, 2852, 2215, 1566, 1528, 1452, 1350, 1296, 1258, 1233, 1163, 1140, 927, 860, 795, 732, 702cm^{-1}.MS-ESI: *m/z* 358.14 [M-H]$^{-}$.

Compound**4k**: Yellowish-brown solid; mp: 250-251°C. ^{1}H NMR ($CDCl_3$): δ = 7.16-6.92 (m, 4H, Ar-H), 4.96 (s, 2H, NH_2), 3.80 (s, 3H, OCH_3), 3.59-3.58 (m, 1H), 3.43-3.40 (m, 1H), 3.26-3.20 (m, 1H), 2.59-2.54 (m, 1H), 2.18 (s, 3H, CH_3), 2.15-2.05 (m, 2H, CH_2), 1.67-1.52 (m, 2H, CH_2). IR (KBr): 3382, 3136, 2921, 2849, 2214, 1610, 1565, 1516, 1450, 1348, 1292, 1252, 1173, 1034, 831, 786cm^{-1}.MS-ESI: *m/z* 342.90 [M-H]$^{-}$.

Compound**4l**: Yellowish-brown solid; mp: 221-222°C. ^{1}H NMR ($CDCl_3$): δ = 7.01-6.69 (m, 3H, Ar-H), 5.06 (s, 2H, NH_2), 3.95 (s, 3H, OCH_3), 3.89 (3H, OCH_3), 3.70-3.67 (m, 1H), 3.52-3.49 (m, 1H), 3.34-3.28 (m, 1H), 2.68-2.62 (m, 1H), 2.25 (3H, CH_3), 2.22-2.12 (m, 2H, CH_2), 1.82-1.59 (m, 2H, CH_2). IR (KBr): 3388, 3150, 2918, 2848, 2213, 1604, 1567, 1517, 1464, 1449, 1409, 1349, 1318, 1296, 1258, 1235, 1136, 1022cm^{-1}.MS-ESI: *m/z* 373.17 [M-H]$^{-}$.

Compound**4m**: Tawny solid; mp: 242-245°C. ^{1}H NMR ($CDCl_3$): δ = 7.36-7.09 (m, 4H, Ar-H), 5.04 (s, 2H, NH_2), 3.65-3.63 (m, 1H), 3.51-3.48 (m, 1H), 3.33-3.28 (m, 1H), 3.01-2.94 (m, 1H, CH), 2.66-2.61 (m, 1H), 2.25 (s, 3H, CH_3), 2.22-2.04 (m, 2H, CH_2), 1.77-1.71 (m, 2H, CH_2), 1.31 (d, 6H, $2CH_3$). IR (KBr): 3397, 3314, 3182, 2967, 2922, 2850, 2796, 2217, 1565, 1451, 1294, 1265, 1252, 1236, 1160, 1138, 1053, 1021, 923, 834, 801, 764, 719cm^{-1}.MS-ESI: *m/z* 355.20 [M-H]$^{-}$.

Compound**4n**: Yellow solid; mp: 287-289°C. ^{1}H NMR ($CDCl_3$): δ = 8.00-7.27 (m, 7H, Ar-H), 5.09 (s, 2H, NH_2), 3.65-3.62 (m, 1H), 3.51-3.48 (m, 1H), 3.37-3.31 (m, 1H), 2.70-2.65 (m, 1H), 2.24 (3H, CH_3), 2.21-2.04 (m, 2H, CH_2), 1.82-1.76 (m, 2H, CH_2). IR (KBr): 3398, 3189, 2922, 2849, 2797, 2213, 1564, 1447, 1297, 1267, 1235, 1161, 1140, 816, 800, 785, 750cm^{-1}.MS-ESI: *m/z* 363.17 [M-H]$^{-}$.

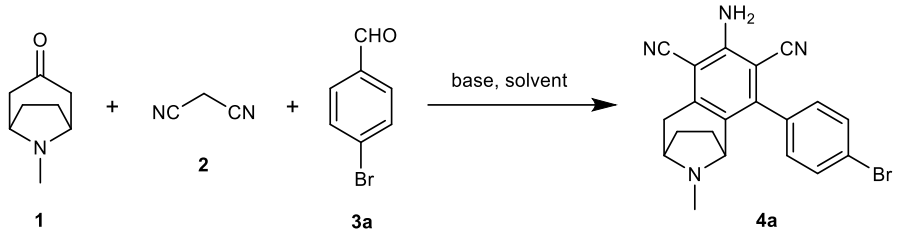
Compound**4o**: Yellow solid; mp: 315-317°C. ^{1}H NMR ($CDCl_3$): δ = 7.53-7.10 (m, 3H, Ar-H), 5.07 (s, 2H, NH_2), 3.80-3.78 (m, 1H), 3.51-3.48 (m, 1H), 3.33-3.27 (m, 1H), 2.65-2.60 (m, 1H), 2.26 (s, 3H, CH_3), 2.23-2.17 (m, 2H, CH_2), 1.82-1.57 (m, 2H, CH_2). IR (KBr): 3389, 3237, 3099, 2923, 2850, 2797, 2216, 1564, 1454, 1297, 1258, 1237, 1161, 1144, 1039, 842, 791, 712cm^{-1}.MS-ESI: *m/z* 319.11 [M-H]$^{-}$.

Results and Discussion

Initially, we explored the three-component reaction oftropinone**1**(0.3 mmol), malononitrile**2**(0.6 mmol) and 4-bromobenzaldehyde **3a** (0.3 mmol) in the presence of 2 equivDBUin toluene at 100°C for 10 hours.The product **4a** was isolated in 30% yield (Table-1, entry 1). Subsequently, other organic bases includingpyrrolidine, piperidine, morpholine, triethylamine, triethylenediamine and sodium methanolate,were used in model reaction to find outthe ideal organic base with the most extraordinary performance. To our disappointment, themodelreaction with other organic bases did not show better results in the perspective ofproductyield (Table-1, entries 2-7).Therefore, DBU was the most prominent organic base and used in the next optimization of process conditions. Then, the reaction solvent was replaced with other organic solvents including ethylene glycol, DMF, DMSO, tetrahydrofuran, acetonitrile and methanol, and the experimental results were shownin Table-1 (entries 8–13). Experimental results indicated that product yieldhas been significantly improved when tetrahydrofuran was used as a solvent. Finally, we turned our attention to study the influences of different amounts of malononitrile, expecting that productyield may increase along with the addition of

more malononitrile(Table-1, entry 11 and entries 14–16). When the amount of malononitrile was doubled, the reaction can proceed smoothly with highest yield. Through systematic screening, we have established a direct and effective means for the preparation of 4-aryl-5,8-epiminobenzo[7]annulenederivative **4a** in 70% yield (Table-1, entry 15).

Table-1: Optimization of process conditions.

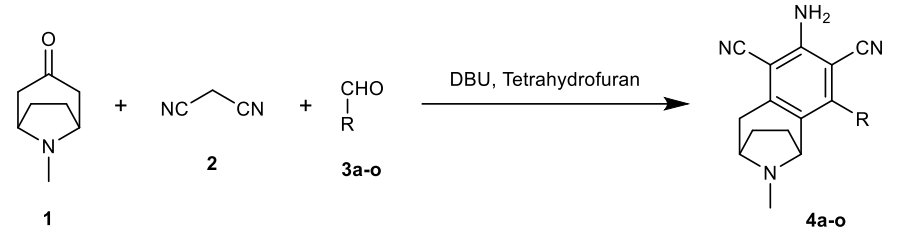


1 3a 4a

Entry	Solvent	Base	Equiv. of malononitrile	Temp (°C)	Yield of 4a[a] (%)
1	Toluene	DBU	2	100	30
2	Toluene	Pyrrolidine	2	100	20
3	Toluene	Piperidine	2	100	27
4	Toluene	Morpholine	2	100	13
5	Toluene	Triethylamine	2	100	18
6	Toluene	Triethylenediamine	2	100	9
7	MeOH	Sodium methanolate	2	60	7
8	Ethylene glycol	DBU	2	100	15
9	DMF	DBU	2	100	14
10	DMSO	DBU	2	100	32
11	Tetrahydrofuran	DBU	2	60	49
12	Acetonitrile	DBU	2	60	28
13	Methanol	DBU	2	60	37
14	Tetrahydrofuran	DBU	3	60	62
15	Tetrahydrofuran	DBU	4	60	70
16	Tetrahydrofuran	DBU	5	60	70

[a] Isolated yield after purification by silica gel column chromatography

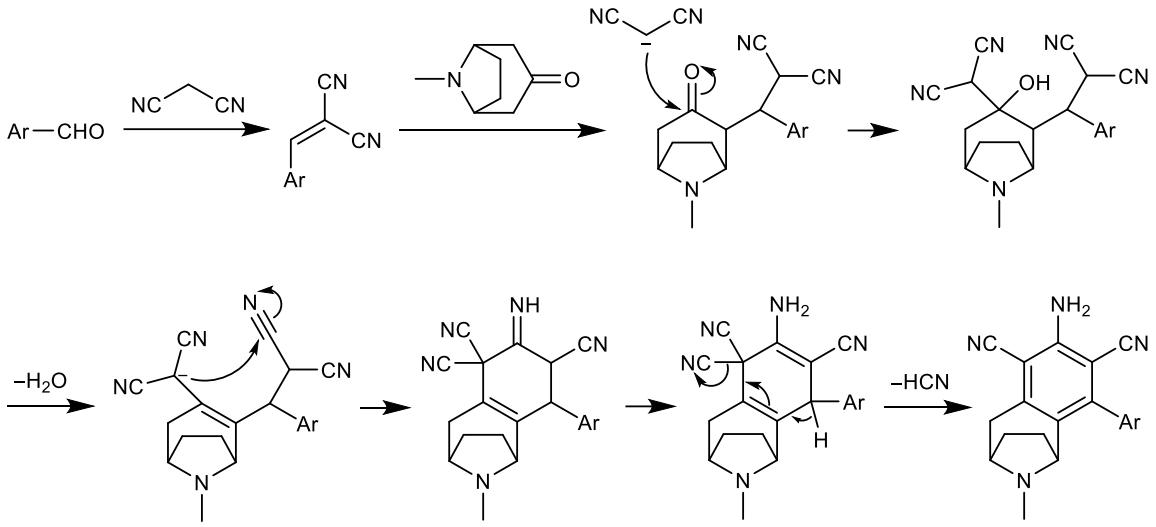
Table-2: Synthesis of 4-aryl-5,8-epiminobenzo[7]annulene derivatives.



1 4a-o

Entry	R (3)	Compound	Yield (%)[a]
1	4-BrC_6H_4 (3a)	4a	70
2	4-FC_6H_4 (3b)	4b	34
3	4-ClC_6H_4 (3c)	4c	44
4	3-ClC_6H_4 (3d)	4d	63
5	4-$CH_3SO_2C_6H_4$ (3e)	4e	30
6	4-$CF_3C_6H_4$ (3f)	4f	39
7	3-$CF_3C_6H_4$ (3g)	4g	51
8	4-CNC_6H_4 (3h)	4h	44
9	4-$NO_2C_6H_4$ (3i)	4i	34
10	3-$NO_2C_6H_4$ (3j)	4j	36
11	4-$CH_3OC_6H_4$ (3k)	4k	42
12	3,4-$(CH_3O)_2C_6H_3$ (3l)	4l	41
13	4-*i*PrC_6H_4 (3m)	4m	70
14	2-Naphthyl (3n)	4n	34
15	2-Thienyl (3o)	4o	35

[a] Isolated yield after purification by column chromatography.



Scheme-1:Reasonablemechanism for the tandem reactions.

Withtheoptimum process conditions at hand,the exploration of substrate scope with regard to the above-mentioned reactionhas become the next most pressing matter(Table-2).Firstly, different aromatic aldehydes containing halogen substituent **3a-d**were used for the reaction, and barely satisfying yields were obtained (Table-2, entries 1-4). Then, various aromatic aldehydes with an electron-withdrawing group **3e-j**were selected as the substrate for this reaction, and the corresponding final products **4e-j**have also been successfully synthesized(Table-2,entries 5-10). Afterwards,thesubstituent group in aromatic aldehydes were replaced by diversified electron-donating groups as substrate (Table-2, entries 11-13), and the reactions proceeded well, affording the target products **4k−m** with acceptable isolated yields (up to 70%). Furthermore, 2-naphthaldehyde **3n**, as a member of fused-ring compounds, was also transformed to the product **4n** smoothly (Table-2, entry 14).To further extend the application of the model reaction, heterocyclic analogue **3o** was employed in this procedure, and the desired targetproduct**4o** was separated successfully (Table-2, entry 15).

A plausible mechanism for the one-pot reaction was outlined in Scheme-1. The reaction started from the Knoevenagel condensation of aromatic aldehydes and malononitrile,and the formed 2-arylidenemalononitrile was then reacted with tropinone through a Michael addition reaction.Subsequently, the keto group in the intermediate product was nucleophilically attacked by another activated malononitrile. Finally, a new benzene ring was successfully constructed by successive domino reactions, including dehydration, intramolecular nucleophilic addition and cyclizationreaction.

Conclusion

On balance, we have exploreda candidate synthesis means for the preparation of potentially biologically active 4-aryl-5,8-epiminobenzo[7]annulenesthroughone-potMichael/cyclization reaction of tropinone, malononitrile and aromatic aldehydes.Under optimized reaction process conditions, various 4-aryl-5,8-epiminobenzo[7]annulenes with different substituents were synthesized in good yield. The tandem domino reaction has broadindustrialapplication prospect in organic and medicinal chemistry due to its notable advantages, such as extensive substrate scope, high atomic utilization efficiency, diminished costs as well as simple and practical operation process.

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